

# Collection, Preparation and Testing of NIST Hard Rock Mine Waste Reference Material SRM 2780

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#### U.S. DEPARTMENT OF THE INTERIOR U.S. GEOLOGICAL SURVEY

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#### Introduction

In the western United States there are an estimated five hundred thousand abandoned mines located on public and private lands. These sites commonly contain significant quantities of mine waste material which was discarded without environmental concern during historic mining operations. Typically this waste is responsible for the generation of acidic (pH 2-4) drainage, containing elevated concentrations of heavy metals. This toxic combination contaminates adjacent waterways leaving them void of biological activity for several kilometers downstream. Remediating historic mine sites often requires millions of public and private dollars and thousands of man hours to return the site to its premining condition.

Accurately evaluating the environmental impact of mine waste is hampered by the variability in laboratory methodology used to assess its hazardous nature. This problem is compounded by the difficulty encountered in comparing data from different labs even though they use the same techniques. These problems have developed in part due to the lack of appropriate geochemical reference materials (GRMs) which mimic mine waste composition. In response to this need, the United States Geological Survey (USGS), the National Institute for Standards and Technology (NIST) and the United States Environmental Protection Agency (U.S. EPA) cooperated in the development of a reference material dedicated to the study of hard rock mine waste. The goal was to produce a GRM which could be used in the development of appropriate analytical procedures, assist in intra-laboratory data evaluation, and help create reliable quality assurance programs for the industry. Due to the complex nature of mine waste drainage, the project was designed to examine both the total element composition and water extractable constituents of the proposed GRM. Determining the total element composition of this material was the responsibility of NIST and the USGS. Quantifying water extractable constituents was coordinated through the Acid Drainage Technology Initiative (ADTI) group. ADTI is a working group of scientists from the public and private sectors dedicated to a better geochemical understanding of coal and hard rock mine waste in the United States.

The USGS was charged with identifying a collection site, as well as collecting and physically preparing two "identical" lots of material. The area selected for sample collection was the Silverton mining district, located in San Juan county ( $392 \text{ mi}^2$ ), Colorado. The district, located near the town of Silverton, Colorado, sits at the southwestern end of the Colorado mineral belt (Cunningham et al., 1995; Ransome, F.L., 1901). The area was extensively mined in the late 1800's to mid 1900's, with limited mining occurring as late as the 1980's. At the height of its mining activity the district officially employed up to 500 people at 31 mines and mills. State production records indicate that between 1873 and 1980, 1.4 billion pounds of lead, copper, and zinc were produced, as well as 51 million ounces of gold and silver (Colorado Bureau of Mines, 1873-1980). The estimated net worth of this activity is placed at 270 billion dollars.

Evidence of this intensive mining activity is still apparent today, in the several abandoned mills located along the Animas River and in the thousands of abandoned mine sites located throughout the valley. Abandoned mine sites are normally characterized by extensive piles of crushed rock adjacent to the mine entrance. These piles range in color from pale yellow to dark red, and typically contain significant quantities of pyrite, galena, arsenopyrite, chalcopyrite, sphalerite and other sulfide minerals. These metal sulfides undergo gradual oxidation producing significant quantities of sulfuric acid. An example of this oxidation process is presented in equations 1-4 for the mineral pyrite ( $FeS_2$ ).

1.  $\text{FeS}_2 + 3.50_2 + H_2 O \rightleftharpoons \text{Fe}^{2+} + 2\text{H}^+ + 2\text{SO}_4^{2-}$ 

2.  $2Fe^{2+} + 0.5O_2 + 2H^+ \rightleftharpoons 2Fe^{3+} + H_2O$ 

3.  $Fe^{3+} + 3H_2O \Rightarrow Fe(OH)_3 + 3H^+$ 

4.  $\text{FeS}_2 + 14\text{Fe}^{3+} + 8\text{H}_2\text{O} \Rightarrow 15\text{Fe}^{2+} + 2\text{SO}_4^{2-} + 16\text{H}^+$ 

The acid generated during pyrite oxidation leaches the surrounding waste material of metals which are then deposited on the surface of adjacent particles as water soluble salts. During high water flow periods such as the spring runoff and summer thundershowers these acidic salts are mobilized from the waste pile, eventually draining into local streams. The combination of elevated metal concentrations and high acidity makes it difficult for indigenous animal species to survive in affected waterways. Extensive federal and private reclamation projects are currently underway in the Silverton area to remediate several of the most important mine waste locations.

The site selected for sample collection was located next to the Mayday mine, which is situated on the northeastern side of the Silverton caldera field, at latitude 37° 50' 52" and longitude 107° 40' 45". The mine sits atop intra-ash flows of andesitic lava, primarily of the tertiary age (Ransome, F.L., 1901; Tweto, O., 1979). Mines located along this side of the Silverton caldera field produced significant quantities of lead, copper and zinc as well as important amounts of tungsten (hubnerite).

#### Sample Collection and Preparation

The Mayday mine site is located on Bureau of Land Management (BLM) land and covers a total area of approximately 1 acre. The waste pile is adjacent to the mine opening and is arranged in a poorly defined two-bench structure. The mine dump is generally devoid of vegetation, despite the fact that the mine site is situated in a lush evergreen forest. Measuring from the adjacent county road, the waste pile is approximately 100 meters high, 200 meters in width, with a sixty degree angle of repose. The mine waste surface material has a gray to pale yellow appearance, with isolated streaks of dark red, brown, or white material. The dump's surface material is stable to foot traffic, and is composed of crushed rock ranging in size from 0.5 to 5 cm in length. Large clods of agglomerated waste material are evident, but are easily disaggregated to particles 1 cm or less in diameter. Erosional effects on the pile are apparent, but areas of deep erosional scaring are limited.

The upper bench of the waste pile was selected for sample collection. It was divided into fifteen sections approximately 10 m in width and 30 m in length (top to the bottom of bench). Sample collection was performed by rapelling down the center of each section in 3 meter increments, and then collecting material within five meters of the rope. Samples were obtained from the top 20 cm of surface material, using a common garden spade. The sample was transferred to five gallon plastic buckets lined with a polyethylene bag. Using this procedure approximately 30 kg of material was collected from each section resulting in a total collected sample weight of over 450 kg.

Upon arrival at the USGS, material from each bucket was transferred to six plastic lined cardboard trays, and dried in a forced air oven (room temperature) for 24 hours. After drying material was mixed for 20 minutes in a 3 ft<sup>3</sup> V-blender, and then transferred back into its original container. A sub-sample from each bucket was obtained using a plastic sample thief (30 cm x 2 cm) and then analyzed for its total element content by the USGS. The contents of selected buckets were eventually combined into one batch and blended for 1 hour using a 10 ft<sup>3</sup> cross flow V-blender. After blending, the sample was transferred back into five gallon containers.

As part of the cooperative arrangement between NIST, USGS, and ADTI, the blended mine waste material was to be divided in two equal size lots. The sample splitting was accomplished using a USGS designed spinning riffler. In this procedure, material from each five gallon container was transferred to a specially designed funnel apparatus which allowed material to flow onto a rotating table (2 m diameter) spinning at approximately six rpms. On the table were positioned a set of 12 stainless steel sample splitters which transferred the material into a set (12) of 5 gallon plastic buckets. After splitting was complete, the odd and even numbered containers were set aside for ADTI and NIST studies respectively. Material for NIST was ground to pass 200 mesh ( $67\mu$ ) and then blended in a 10 ft<sup>3</sup> cube blender for 24 hours. The ground material was transferred to 5 kg polyethylene bags, boxed, and radiation sterilized at COBE Laboratories, Lakewood, Colorado. The radiation dosage during sterilization ranged between 1.5 and 3.2 Mrad.

In the final preparation step, the ground/sterilized material was reblended and then split using the spinning riffler to produce ~4 kg aliquots. The spinning riffler was refitted with a 72 position sample ring, and each 4 kg aliquot was used to fill containers(55-60g) with material. A total of 2200 units were prepared in this manner.

# Results and Discussion

Prior to shipping the bottled material to NIST, the USGS samples for major and trace elements to determine analyzed between-bottle homogeneity. Analytical methods used included Wavelength Dispersive X-ray Fluorescence (WD-XRF), Inductively Coupled Plasma Atomic Emission Spectroscopy (ICP-AES), Instrumental Neutron Activation Analysis (INAA), Hydride Generation Atomic Absorption Spectroscopy (HYG-AAS), Cold Vapor Atomic Absorption Spectroscopy (CV-AAS), Combustion Infrared detection (C-ID), Coulometric Titration (CT), Fire Assay (FA), Graphite Furnace Atomic Absorption Spectroscopy (GF-AAS), and Potentiometric Titration (PT). Analyses were performed at the USGS laboratory facilities, Lakewood, Colorado and at the USGS contract lab in Ottawa, Canada. Contract lab results are designed to provide information on those elements not analyzed at the USGS. Samples used in the USGS homogeneity test represented every 100<sup>th</sup> bottle (22), as well as six samples selected at random. A sample of NIST SRM 2710 was included in the sample set for quality control purposes. All samples were analyzed on an as received basis. Α brief description of USGS techniques used and their acceptable average element variation is presented in Table 1. More detailed information on USGS analytical procedures is available (Arbogast 1996). Homogeneity testing performed at the USGS contract lab used six samples selected at random. Analytical procedures used at the USGS contract lab are similar to USGS procedures with only minor modifications.

Results from the between-bottle analysis of SRM 2780 are presented in appendices A through E and summarized in table 2 for both the USGS and contract lab. Results are presented by individual techniques except for CV-AAS, HY-AAS, and C-ID which are grouped under the general heading LATA. In this study the estimate of total analytical uncertainty is represented by the percent relative standard deviation (%RSD) as determined using equation 5.

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The %RSD represents a combination of two error components. The first reflects the between-bottle variation in element concentration (Vconc) and the second reflects the error associated with the actual analytical measurement (Em). The USGS value of Em for any technique is estimated through the USGS method validation process, which relies on the analysis of selected geochemical reference materials over three nonconsecutive days. Table 1 presents information on the average %RSD for each USGS technique used.

For the majority of elements quantified in SRM 2780, the observed %RSD is equal to or less than the technique Em. Under USGS guidelines, the material is considered homogeneous for that element within the limits established by the method. Selected elements display %RSD in excess of the Em values and should be carefully evaluated before they are considered homogeneously distributed. These elements include Na, P (WDXRF), Au, Ni, W (INAA), Hg (CV-AAS), and Co(ICP-AES).

Examination of results in table 2 reveal that several elements are present at elevated concentrations. In the case of Ag, As, Cd, Pb, Sb, and Zn, element concentrations are at least two orders of magnitude greater than average crustal material (Fortescue, J., 1992; Schacklette, H. and Boerngen, J., 1984). Concentrations of other elements such as Cu, S, and W, are 2 to 10 times greater than average abundances. USGS X-ray diffraction studies indicate that the major mineral phases in the sample include, quartz, muscovite, feldspar, chlorite, a series of jarosite minerals as well as a significant fraction of amorphous alumino silicates. Also present are minor amounts of galena, sphalerite, and pyrite.

An evaluation of analytical accuracy was performed by comparing USGS results for the analysis of SRM 2710 with NIST certificate values (table 3). NIST SRM 2710 was selected as a QC sample because its element concentrations and mineralized composition are similar to SRM 2780. Prior to sample analysis a confidence interval was established for each element based on

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certificate information and USGS quality control guidelines. In the case of NIST certified elements, the acceptable element concentration range (AECR) was established as the certified value  $\pm$  twice the NIST 95% confidence interval. Elements with NIST information values were assigned an AECR equal to the recommended value  $\pm$  20% of the mean value. Element concentrations in table 3 that fall outside their calculated AECR are contained within parentheses. All values were corrected for their percent moisture values (1.6%), which was previously determined by the USGS.

Examination of major element results in table 3 reveal that the majority (82%) of major elements concentrations fall within their designated AECR. Exceptions are noted for Na, Si (WDXRF), Ti (ICP-AES), and Ca (INAA). The bias observed for Na and Si using WDXRF reflects a consistent trend in USGS analysis of this SRM (Wilson et al., 1994). Comparison of USGS results with NIST certificate values for minor and trace elements show that 88% of elements have observed concentrations within their calculated AECR. Exceptions to this trend include Co, Cr, Ga,(ICP-AES), and Ag, Cr, Ni, Zn, (INAA).

#### Conclusion

The NIST hard rock mine waste reference material SRM 2780 was evaluated for element homogeneity by the USGS using a variety of analytical procedures. Results indicate that for the majority of elements tested the total element concentrations show no significant between-bottles variation. Concentrations of selected metals are several orders of magnitude above normal crustal abundances and major mineralogical phases are consistent with the mineralized nature of the area.

#### Bibliography

Arbogast, B.F., 1996, Analytical Methods Manual for the Mineral Resources Surveys Program, U.S. Geological Survey, Open File report 96-525, p 248

Colorado Bureau of Mines, 1898-1985, A Summary of Mineral Industry Activities in Colorado, State of Colorado, p 100.

Cunningham, C.G., Naeser, C.W., Marvin, R.F., Luedke, R.G., Wallace, A.R., 1994, Ages of Selected Intrusive Rocks and Associated Ore Deposits in the Colorado Mineral Belt, U.S. Geological Survey Bulletin, 2109, 26 p.

Fortescue, J., 1992, Landscape geochemistry: retrospect and prospect — 1990: Applied Geochemistry, v 7, p. 1-53.

Ransome, F.L., 1901, A Report on the Economic Geology of the Silverton Quadrangle, Colorado, U.S. Geological Survey Bulletin, p262.

Schacklette, H.T., and Boerngen, J., 1984, Element concentrations in soils and other surficial materials of the conterminous United States, U.S. Geological Survey Professional Paper 1270, p 105.

Tweto, O., 1979, Geologic Map of Colorado 1:500,000, U.S. Geological Survey Map GSG-6-2.

Table 1.	USGS a	nalytical	procedures
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Method	Decomposition Procedure	Sample wt g	Precision %RSD
ICP-AES	multi acid	0.200	5 - 10
WD-XRF	Lithium Tetaborate Fusion	0.800	1 - 2
INAA	none	0.7	1 - 10
CV-AAS	Dichromate/ nitric acid	0.25	2 - 5
HYG-AAS	multi acid	0.25	5 - 10
C-ID	High temp. combustion	0.25	5 - 10

		USGS			USGS			USGS			USGS				
		WDXRF			<b>ICP-AES</b>			INAA			LATA		Co	ntract	lab 🛛
Element	AVG	STDS	%RSD	AVG	STDS	%RSD	AVG	STDS	%RSD	AVG	STDS	%RSD	AVG	STDS	%RSD
AI %	8.79	0.03	0.30	8.50	0.23	2.7									
Ctot										0.17	0.01	6.0	0.17	0.02	12
Cinorg													0.010	0.004	10
Corg		0.004											0.16	0.02	12
Ca	0.20	0.004	2.1	0.19	0.01	4.5	0.29	0.01	3.4						
Fe FeO	2.73	0.02	0.6	2.7	0.02	0.8	2.77	0.03	1.2				0.68	0.04	6
K	3.43	0.03	0.74	3.2	0.09	2.9	4.08	0.17	4.1				0.00	0.04	°
Mg	0.54	0.05	1.2	0.51	0.03	1.0	4.00	0.17							
Na	(0.32)	0.01	3.1	0.21	0.01	3.3	0.22	0.002	0.81						
P	(0.062)	0.003	5.3	0.04	0.00	6.9	0.22	0.002	0.01						[
Stot	(		0.0		0.00	•.•				1.23	0.02	1.2	1.29	0.03	2
Si	34.0	0.06	0.2								0.02				-
Ti	0.698	0.004	0.6	0.64	0.04	6.3									
Ag ppm				27	1	2.6	22.6	0.6	2.7						
As				72	6	7.9	48.2	0.8	1.7	44	3	7.8			
Au				<8	-	•	(0.184)	26	14				0.20	0.04	18
Ba				900	40	4.5	964	17	1.8						
Cd				12	1	4.1	12.0	1.0	8.5						
Co				61	5	7.9	67.1	0.9	1.3						
Co Cr				(2.1)	0.5	24	2.21	0.04	1.8						
Cr				48	2	3.7	40.6	1.4	3.5						
Cu				220	11	5.2	13.1	0.1	1.0						
Ga				26.0	0.5	1.8									
Hf				20.0	0.5	1.0	4.41	0.07	1.5						
Hg								0.07	1.5	(0.73)	0.10	14	0.7	0.1	14
Но				<4	-	-	0.84	0.03	3.9	(00)	0.10	••	•		
La				38	3	9.0	37.6	0.4	0.9						
Li				18	0.2	1.1									
Mn	(450)	30	7.2	450	10	2.3									
Мо				(13)	2.6	20	10.6	0.9	8.9						
Nb				18	1	5.3									
Nd				26	2	7.4	30.5	0.6	2.1						
Ni				12	0.5	4.0	(19.5)	3.2	16						
Pb				5700	110	1.9		-							
Rb							173	2	1.2				405	10	
Sb Sc				23	2	7.5	159 23.7	2 0.3	1.3				125	10	8
Se				23	2	7.5	3.4	0.3	1.1 5.2	5.15	0.13	2.5	4.4	0.4	9
Sr				220	6	2.7	294	0.2 6	5.z 2.1	5.15	0.15	2.0	<b>.</b>	0.4	3
Tb	1			220	5	2.1	0.58	0.01	1.8						
Te							0.00	0.01	1.0				5.3	0.7	13
Th	ł			(8)	2	32	11.5	0.1	1.3				9.5	0.5	6
П				(-)	-								4.9	0.4	7
Tm				-			0.40	0.02	3.9	1					
U.				<100	-	-	3.95	0.08	2.0						
V				260	5	2.0									
W							(20)	3	16				23.8	2.1	9
Zn				2800	130	4.7	2380	53	2.2						
Zr	L						162	16	9.9	L					

### Table 2. Summary results for USGS and contract lab analysis of NIST SRM 2780

Element concentration in paraentheses have observed %RSD in excess of method %RSD

Element	WDXRF	ICP-AES	INAA	LATA	value	+/-	limit	limit
AI %	6.45	6.50			6.44	0.08	6.6	6.28
Ctot				2.99	3		3.6	2.4
Ca	1.26	1.22	(1.82)		1.25	0.03	1.31	1.19
Fe	3.31	3.35	3.47		3.38	0.1	3.58	3.18
ĸ	2.07	2.03	2.08		2.11	0.11	2.33	1.89
Mg	0.86	0.86			0.853	0.042	0.937	0.769
Na	(1.32)	1.12	1.11		1.14	0.06	1.26	1.02
Р	0.12	0.11			0.106	0.015	0.136	0.076
Stot				0.02				
Si	(28.31)				28.97	0.18	29.33	28.61
Tì	0.27	0.13			0.283	0.01	0.303	0.263
Ag ppm		36	(28)		35.3	1.5	38.3	32.3
As		650	668	599	626	38	702	550
Au		<8	0.49		0.6		0.7	0.5
Ba		691	740 _		707	51	809	605
Cd		21	21		21.8	0.2	22.2	21.4
Ce		58	57.3		57		68	46
Co		(13)	9.11		10		12	8
Cr		(30)	(30.3)		39		47	31
Cs			109					
Cu		3048			2950	130	3210	2690
Ga		(16)			34	,	41	27
Hf			3.58		3.2		3.8	2.6
Hg				30	32.6	1.8	36.2	29
Ho			0.89		0.6		0.7	0.5
La		35.6	30.7		34		41	27
Li		37.6						
Mn	9956.80	9550			10100	400	10900	9300
Мо		20.3	24.2		19		23	15
Nb		16.3						
Nd		21.3	25.9		23		28	18
Ni		16	(10.1)		14.3	1	16.3	12.3
Pb		5588	()		5532	80	5692	5372
Rb			129		120		144	96
Sb			36		38.4	3	44.4	32.4
Sc		10	9.1		8.7	5	10.4	7.0
Se			0.28	0.55	0.7		10.4	7.0
Sr		345	358	0.00	330		396	264
		040	0.68		330		290	204
Tb Th			13		13		16	10
Tm			0.37		13		16	10
um U			27		25		30	20
V		77	21		25 76.6	22		20 72
			91			2.3	81.2	
W 7=		6000		į	93	04	112	74 6770
Zn Zr		6909	(6532)		6952	91	7134	6770
Zr			111		L			

Table 3. USGS summary results for the analysis of NIST SRM 2710\*

\* values corrected for percent moisture concentrations in parentheses are outside AECR for SRM 2710

0	30.0	0.065	0.31	0.05	0.54	3.41	2.72	0.21	8.84	F-027742
0	30.0	0.065	0.34	0.05	0.54	3.40	2.74	0.20	8.79	F-027741
<u>.</u>	30.0	0.065	0.32	0.05	0.54	3.42	2.72	0.21	8.79	F-027740
0.6	30.0	0.057	0.33	0.05	0.54	3.42	2.72	0.21	8.79	F-027739
0	30.1	0.065	0.33	0.05	0.54	3.45	2.75	0.20	8.84	F-027738
0.6	30.0	0.057	0.32	0.05	0.53	3.45	2.73	0.20	8.79	F-027737
0.6	30.0	0.065	0.33	0.05	0.54	3.45	2.76	0.20	8.79	F-027736
0.6	30.0	0.065	0.33	0.04	0.54	3.45	2.73	0.20	8.79	F-027735
0.7	30.1	0.065	0.33	0.05	0.55	3.46	2.74	0.21	8.79	F-027734
0.7	30.0	0.061	0.32	0.05	0.54	3.45	2.71	0.21	8.79	F-027733
0.7	30.1	0.070	0.32	0.04	0.54	3.46	2.73	0.21	8.79	F-027732
0.707	30.1	0.061	0.33	0.05	0.53	3.40	2.72	0.21	8.79	F-027731
0.6	30.0	0.061	0.32	0.05	0.54	3.39	2.73	0.21	8.79	F-027730
0.7	30.0	0.061	0.31	0.05	0.55	3.40	2.72	0.21	8.79	F-027729
0.7	29.9	0.061	0.31	0.05	0.53	3.39	2.69	0.20	8.73	F-027728
0.6	30.0	0.061	0.33	0.05	0.52	3.40	2.71	0.20	8.79	F-027727
0.7	30.0	0.057	0.33	0.04	0.54	3.42	2.72	0.20	8.79	F-027726
0.6	30.0	0.061	0.33	0.05	0.54	3.41	2.75	0.21	8.73	F-027725
0.7	30.0	0.061	0.33	0.05	0.52	3.45	2.71	0.20	8.79	F-027724
0.7	30.0	0.065	0.33	0.04	0.54	3.45	2.72	0.20	8.84	F-027723
0.6	29.9	0.065	0.31	0.05	0.54	3.44	2.73	0.21	8.79	F-027722
0.6	30.0	0.061	0.34	0.04	0.53	3.45	2.71	0.20	8.79	F-027721
0.6	30.0	0.061	0.31	0.05	0.54	3.45	2.73	0.21	8.84	F-027720
0.6	30.0	0.061	0.31	0.05	0.53	3.45	2.75	0.20	8.79	F-027719
=	SI %	Р%	Na %	Mn %	Mg %	K %	Fe, %	Ca%	Al %	Lab No

Appendix A. WD-XRF results for NIST SRM 2780

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Appendix B. ICP-AES results for NIST SRM 2780

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F-027742	F-027741	F-027740	F-027739	F-027738	F-027737	F-027736	F-027735	F-027734	F-027733	F-027732	F-02//31	-021130	E 007730	F-027720	F-027728	F-027727	F-027726	F-027725	F-027724	F-027723	F-027722	F-027721	F-027720	F-027719	lab no	741170-1	E-027742	F-027741	F-027740	F-027739	F-027738	F-027737	F-027736	F-027735	F-027734	F-027733	F-027732	F-027731	F-027730	F-027729	F-027728	F-027727	F-027726	F-027725	F-027724	F-027723	F-027722	F-027721	F-027720	Lab no	Appendix C.
37.4	37.4	37.5	37.5	<b>3</b> 8.1	37.7	37.3	37.8	38.2	37.3	37.8	37.9	27.1	37.0	37.0	37.9	37.7	36.9	37.8	37.9	37.4	37.8	37.9	37.7	37.0	La ppm		3 !	277	2.79	2.80	2.78	2.77	2.76	2.76	2.74	2.76	2.76	2.71	2 79	2.73	2.81	277	2.73	2.83	200	278	279	2 80	2.78	5 75	
11.30	9.86	10.70	9.80	13.40	9.98	12.80	9.30	12.50	11.40	9.71	10.30	40 20 9. 12	5.0	11 10	9.53	8.52	10.20	10.30	10.60	12.70	10.10	9.71	9.92	10.50	Mo ppm	0.02	0.01	0.37	0.34	0.32	0.36	0.31	0.30	0.35	0.31	0.28	0.29	0.26	0.27	0.26	0.26	0.25	0.27	0.27	0 C C	0.27	0.27	0.27	0.25	Ca %	INAA resu
31.2	30.9	<b>30.0</b>	30.6	31.5	31.1	29.9	29.8	31.7	30.1	30.3	31.2	0.67		2 5 л	31.3	31.4	29.6	31.0	29.9	30.5	30.2	30.9	29.7	29.8	Nd ppm	0.22	3	0 22	0.22	0.21	0.22	0.22	0.22	0.22	0.23	0.22	0.22	022	0 22	0.22	0.22	023	0.22		2	2	25	0 23			INAA results for NIST SRM 2780
19.6	20.1	18.5	16.9	18.8	16.5	17.1	14.4	18.6	16.5	23.2	16.3	17.0	1 C.	18 1	23.7	25.2	21.6	17.0	18.2	17.5	23.3	20.3	25.1	24.1	Ni ppm	0.03	3 20	4.08	4.8	4.65	4.25	4.52	3.77	4.32	4.35	4.52	4.44	4.09	3.96	3.94	3.92	3.97	3.94	3.87	4 25	3 78	406	406	3.77	a ⊼ %	SRM 278
173	174	174	173	175	174	172	173	170	171	173	168	173	173	173	173	172	170	176	175	173	172	176	173	173	Rb ppm	20.02	20.8	227	22.9	21.9	23.3	23.0	12.4	22.8	22.2	22.2	22.8	222	23.4	23.6	23.4	22.0	21.6	22.4	30	21	2	22.4	21	Ag ppm	
155	160	158	161	160	157	156	156	155	157	158		į	1 8	180	161	158	156	162	161	161	<b>1</b> 60	162	159	157	Sb ppm	ł.	48.0	46.0	48.4	48.0	<b>5</b> 0.8	47.7	45.9	47.9	47.4	49.7	46.4	49.0	48.6	49.8	48.7	48.2	47.8	47.7	48 7	47 3	48.6	48 5	49.0	As ppm	
23.7	23.3	23.9	23.6	24.0	23.9	23.7	23.7	23.6	23.7	23.8	23.5	0 0 1 0	2 2 2 0 0	22.0	235	23.8	23.4	24.1	24.2	23.8	24.0	24.2	23.7	23.4	Sc ppm	1,0	176	135	209	201	203	557	199	133	181	119	138	<u>1</u> ස	204	206	145	171	<u>1</u> ස	<b>1</b> 4	153	173	5 5 5	212	ਤੀ ਹ	Au ppb	
282	299	291	288	<b>30</b> 8	314	317	307	290	290	287	293	167	202	208	307	286	293	278	294	287	283	288	290	288	Sr ppm	900	о Л Л	970	90 200	965	1000	965	9 <b>9</b> 1	950	966	948	943	933	961	958	973	977	936	967	95 57	53	866 200	983	952	Ba ppm	
0.60	0.58	0.58	0.57	0.60	0.57	0.59	0.58	0.58	0.59	0.57	0.56	0.09		о ла ла	0.58	0.57	0.58	0.59	0.57	0.58	0.58	0.58	0.55	0.56	Tb ppm	12.0	<b>ј</b>	12.5	10.7	13.7	13.7	11.9	12.8	12.8	10.1	10.1	11.3	11.1	11.4	13.3	12.0	12.7	11.5	6.6	10.9	13.0	12.8	11.9	13.8	Cd ppm	
11.4	11 <u>.</u> 3	11.6	11.6	11.6	11.6	11.5	11.4	11.4	11.4	11.4	11.2	11	1 - n 4	11 2	11 5 5	11.4	11.3	11.6	11.7	11.4	11.6	11.6	11.3	11.2	Th ppm	07.1	67 1	66.2	66.8	67.1	68.0	67.2	<del>6</del> 6.1	66.7	68.5	67.2	66.9	66.3	68.8	65.9	67.2	66.9	65,9	68.1	67 8	67.9	67.5	66.7	66.9	Ce ppm	1
0.41	0.41	0.40	0.39	0.39	0.38	0.37	0.39	0.41	0.40	0.39	0.38	0.41		0.43	0.41	0.40	0.39	0.37	0.42	0.39	0.42	0.39	0.39	0.41	Tm ppm	<u>•</u>	3 <b>1</b> 3	2 25	2.17	2.26	2.21	2.17	2.18	2.17	2.14	2.18	222	2.18	2.21	2.23	2.21	2.23	2.16	222	2.33	2.18	2.19	2.24	2.26	Co ppm	1
3.97	4.00	3.94	3.92	3.98	3.99	3.91	3.97	3.92	3.93	3.88	3.78	) 19:01 19:01		4 ?	4.07	390	3.95	4.04	4.03	3.90	3.98	4.01	3.92	3.93	U ppm	10.1	45 7	41.0	41.6	42.8	41.5	39.5	39.2	<b>44</b> .2	38.5	40.7	40 <u>.</u> 4	37.4	39.2	<b>39</b> .8	42.5	40.7	<b>4</b> 0,0	43.0	40.5	39.4	41 i	42.0	8 	Cr ppm	)
15.8	21.2	34.8	18.6	18.4	14.2	20.7	22.8	18.5	19.0	19.9	28.1	20.0	) ) ) ) ) )	18.7	19.8	14.8	17.2	16.2	12.7	17.5	21.1	16.8	22.2	20.1	W ppm	10.0	130	12.9	13.2	13.2	13.2	13.1	13.0	13.0	12.8	13.1	13.2	12.9	13.2	13.1	13.2	13.2	13.0	13.3	13.3	13.3	13.1	13.2	13.0	Cs ppm	)
2380	2320	2380	2400	2380	2270	2290	2390	2390	2340	2500	2410	2320	32	2240	2450	2460	2310	2430	2430	2350	2370	2330	2410	2430	Zn ppm	0.00	3 60	3.63	3.55	3.55	3.61	3.57	3.66	3.54	3.53	3.57	3.64	3.55	3.60	3.58	3.57	3.50	3.51	3.S	3.59	3.56	3.62	3.66	3.58	Gd ppm	<u>)</u>
170	164	170	153	171	<b>1</b> 49	149	159	177	159	172	16/	2 2 8	130	1	159	143	<b>1</b> 65	<b>1</b> 65	137	152	183	195	172	8	Zr ppm	7.05	4 30	4.45	4.46	4.41	4.56	4.41	4.39	4.45	4.34	4.41	<u>4</u> .45	4.27	4.40	4.37	4.47	4.42	4.33	4.51	4.45	4 29	4.41	4.47	4. 4 <del>0</del>	4 40	

F-027742	F-027741	F-027740	F-027739	F-027738	F-027737	F-027736	F-027735	F-027734	F-027733	F-027732	F-027731	F-027730	F-027729	F-027728	F-027727	F-027726	F-027725	F-027724	F-027723	F-027722	F-027721	F-027720	F-027719	Lab No	Appendix D.
44	42	47	47	42	46	44	44	46	47	43	38	42	5	41	45	38	42	42	38	45	40	41	49	As, ppm	USGS
4.8	4.9	5.2	5.1	5.1	5.1	5.1	5.1	5.1	5.2	5.2	5.2	5.2	5.3	5.0	5.3	5.2	5.0	5.2	5.2	5.1	5.3	5.3		Se, ppm	single elen
0.65	0.72	0.70	0.69	0.72	0.65	0.72	0.67	0.72	0.66	0.79	0.67	0.84	0.70	0.65	0.73	0.69	0.68	0.90	0.87	0.69	- -	0.63	0.72	Hg, ppm	single element results for NIST
0.19	0.17	0.16	0.17	0.17	0.18	0.17	0.17	0.17	0.17	0.17	0.17	0.16	0.17	0.16	0.17	0.17	0.17	0.17	0.2	0.17	0.17	0.17	0.2	Ctot %	
1.24	1.21	1.23	1.22	1.23	1.23	1.21	1.24	1.21	1.24	1.21	1.22	1.21	1.21	1.22	1.21	1.23	1.24	1.23	1.23	1.24	1.24	1.26	1.26	Stot %	SRM 2780

USGS single element results for NIST SRM 2780

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Appendix E.		Contract la	ab single o	Contract lab single element results for NIST SRM 2780	ults for NIS	ST SRM 27	80						
	As ppm	Au ppm	Cco3	Corg %	Ctot %	FeO %	Hg ppm	Sb ppm	Se ppm	Stot %	Te ppm	TI ppm	W ppm
_	(5)	(4)	(2)		<u>(</u>	(2)	(3)	(5)	(5)	(1)	(6)	(6)	(7)
	48.8	0.261	0.01	0.18	0.19	0.6	0.73	129	4.8	1.32	5.9	4.6	25
•	54.2	0.199	0.01	0.15	0.15	0.7	0.69	130	4.4	1.3	4.5	4.7	26
C-115703	55	0.217	0.01	0.15	0.16	0.7	0.76	129	4.0	1.31	5.6	5.3	24
	53.5	0.188	0.01	0.19	0.20	0.7	0.51	133	4.6	1.24	5.6	4.6	25
-	48.1	0.152	0.01	0.16	0.17	0.7	0.66	106	4.5	1.28	5,9	4.6	23
	50.8	0.199	0.01	0.14	0.15	0.7	0.79	121	3.8	1.31	4.4	5. <b>3</b>	20

1 LECO Comb IR 2 Potentionmetric titration 3 Cold Vapor AAS 4 Fire Assay

5 Hydride generation AAS 6 Graphite furnace-AAS 7 INAA