

A UNITED STATES
DEPARTMENT OF
COMMERCE
PUBLICATION



A11106 360246

NBS SPECIAL PUBLICATION 260-43

Standard Reference Materials:

PREPARATION AND HOMOGENEITY CHARACTERIZATION OF AN AUSTENITIC IRON—CHROMIUM—NICKEL ALLOY

QC

100

.U57

no.260-43

1972

c.2

U.S.
DEPARTMENT
OF
COMMERCE

National
Bureau
of
Standards

NATIONAL BUREAU OF STANDARDS

The National Bureau of Standards¹ was established by an act of Congress March 3, 1901. The Bureau's overall goal is to strengthen and advance the Nation's science and technology and facilitate their effective application for public benefit. To this end, the Bureau conducts research and provides: (1) a basis for the Nation's physical measurement system, (2) scientific and technological services for industry and government, (3) a technical basis for equity in trade, and (4) technical services to promote public safety. The Bureau consists of the Institute for Basic Standards, the Institute for Materials Research, the Institute for Applied Technology, the Center for Computer Sciences and Technology, and the Office for Information Programs.

THE INSTITUTE FOR BASIC STANDARDS provides the central basis within the United States of a complete and consistent system of physical measurement; coordinates that system with measurement systems of other nations; and furnishes essential services leading to accurate and uniform physical measurements throughout the Nation's scientific community, industry, and commerce. The Institute consists of a Center for Radiation Research, an Office of Measurement Services and the following divisions:

Applied Mathematics—Electricity—Heat—Mechanics—Optical Physics—Linac Radiation²—Nuclear Radiation²—Applied Radiation²—Quantum Electronics³—Electromagnetics³—Time and Frequency³—Laboratory Astrophysics³—Cryogenics³.

THE INSTITUTE FOR MATERIALS RESEARCH conducts materials research leading to improved methods of measurement, standards, and data on the properties of well-characterized materials needed by industry, commerce, educational institutions, and Government; provides advisory and research services to other Government agencies; and develops, produces, and distributes standard reference materials. The Institute consists of the Office of Standard Reference Materials and the following divisions:

Analytical Chemistry—Polymers—Metallurgy—Inorganic Materials—Reactor Radiation—Physical Chemistry.

THE INSTITUTE FOR APPLIED TECHNOLOGY provides technical services to promote the use of available technology and to facilitate technological innovation in industry and Government; cooperates with public and private organizations leading to the development of technological standards (including mandatory safety standards), codes and methods of test; and provides technical advice and services to Government agencies upon request. The Institute also monitors NBS engineering standards activities and provides liaison between NBS and national and international engineering standards bodies. The Institute consists of a Center for Building Technology and the following divisions and offices:

Engineering Standards Services—Weights and Measures—Invention and Innovation—Product Evaluation Technology—Electronic Technology—Technical Analysis—Measurement Engineering—Fire Technology—Housing Technology⁴—Federal Building Technology⁴—Building Standards and Codes Services⁴—Building Environment⁴—Structures, Materials and Life Safety⁴—Technical Evaluation and Application⁴.

THE CENTER FOR COMPUTER SCIENCES AND TECHNOLOGY conducts research and provides technical services designed to aid Government agencies in improving cost effectiveness in the conduct of their programs through the selection, acquisition, and effective utilization of automatic data processing equipment; and serves as the principal focus within the executive branch for the development of Federal standards for automatic data processing equipment, techniques, and computer languages. The Center consists of the following offices and divisions:

Information Processing Standards—Computer Information—Computer Services—Systems Development—Information Processing Technology.

THE OFFICE FOR INFORMATION PROGRAMS promotes optimum dissemination and accessibility of scientific information generated within NBS and other agencies of the Federal Government; promotes the development of the National Standard Reference Data System and a system of information analysis centers dealing with the broader aspects of the National Measurement System; provides appropriate services to ensure that the NBS staff has optimum accessibility to the scientific information of the world, and directs the public information activities of the Bureau. The Office consists of the following organizational units:

Office of Standard Reference Data—Office of Technical Information and Publications—Library—Office of International Relations.

¹ Headquarters and Laboratories at Gaithersburg, Maryland, unless otherwise noted; mailing address Washington, D.C. 20234.

² Part of the Center for Radiation Research.

³ Located at Boulder, Colorado 80302.

⁴ Part of the Center for Building Technology.

APR 29 1974

Standard Reference Materials:

Preparation and Homogeneity Characterization of an Austenitic Iron–Chromium–Nickel Alloy

Harvey Yakowitz and Arthur W. Ruff
Metallurgy Division

and

Robert E. Michaelis
Office of Standard Reference Materials

Institute for Materials Research
National Bureau of Standards
Washington, D.C. 20234



U.S. DEPARTMENT OF COMMERCE, Peter G. Peterson, *Secretary*
NATIONAL BUREAU OF STANDARDS, Lawrence M. Kushner, *Acting Director*,

Issued November 1972

Library of Congress Catalog Card Number: 72-600 3 12

National Bureau of Standards Special Publication 260-43

Nat. Bur. Stand. (U.S.), Spec. Publ. 260-43, 19 pages (Nov. 1972)

CODEN: XNBSAV

PREFACE

Standard Reference Materials (SRM's) as defined by the National Bureau of Standards are "well-characterized materials, produced in quantity, that calibrate a measurement system to assure compatibility of measurement in the nation." SRM's are widely used as primary standards in many diverse fields in science, industry, and technology, both within the United States and throughout the world. In many industries traceability of their quality control process to the national measurement system is carried out through the mechanism and use of SRM's. For many of the nation's scientists and technologists it is therefore of more than passing interest to know the details of the measurements made at NBS in arriving at the certified values of the SRM's produced. An NBS series of papers, of which this publication is a member, called the NBS Special Publication - 260 Series is reserved for this purpose.

This 260 Series is dedicated to the dissemination of information on all phases of the preparation, measurement, and certification of NBS-SRM's. In general, much more detail will be found in these papers than is generally allowed, or desirable, in scientific journal articles. This enables the user to assess the validity and accuracy of the measurement processes employed, to judge the statistical analysis, and to learn details of techniques and methods utilized for work entailing the greatest care and accuracy. It is also hoped that these papers will provide sufficient additional information not found on the certificate so that new applications in diverse fields not foreseen at the time the SRM was originally issued will be sought and found.

Inquiries concerning the technical content of this paper should be directed to the author(s). Other questions concerned with the availability, delivery, price, and so forth will receive prompt attention from:

Office of Standard Reference Materials
National Bureau of Standards
Washington, D.C. 20234

J. Paul Cali, Chief
Office of Standard Reference Materials

CONTENTS

	PAGE
I. INTRODUCTION.....	2
II. PREPARATION OF THE ALLOY.....	2
III. PREPARATION FOR MICROPROBE ANALYSIS.....	3
IV. HOMOGENEITY CHARACTERIZATION.....	4
V. ELECTRON PROBE MICROANALYSIS.....	6
VI. CONCLUSIONS.....	7
VII. REFERENCES.....	10

LIST OF TABLES

I. Electron Probe Microanalysis Results for Fe, Cr and Ni in SRM 479. Standards were pure Fe, Cr and Ni. True concentrations in weight fractions: Fe = 0.710; Cr = 0.183; Ni = 0.017.....	9
---	---

LIST OF FIGURES

I. Microstructure observed on a transverse surface after polishing and etching. Magnification 160X.....	11
---	----

OTHER NBS PUBLICATIONS IN THIS SERIES

- NBS Spec. Publ. 260, Catalog of Standard Reference Materials, July 1970. 75 cents.* (Supersedes NBS Misc. Publ. 260, January 1968 and NBS Misc. Publ. 241, March 1962.)
- NBS Misc. Publ. 260-1, Standard Reference Materials: Preparation of NBS White Cast Iron Spectrochemical Standards, June 1964. 30 cents.*
- NBS Misc. Publ. 260-2, Standard Reference Materials: Preparation of NBS Copper-Base Spectrochemical Standards, October 1964. 35 cents.*
- NBS Misc. Publ. 260-3, Standard Reference Materials: Metallographic Characterization of an NBS Spectrometric Low-Alloy Steel Standard, October 1964. 20 cents.* (Out of print).
- NBS Misc. Publ. 260-4, Standard Reference Materials: Sources of Information on Standard Reference Materials, February 1965. 20 cents.* (Out of print).
- NBS Misc. Publ. 260-5, Standard Reference Materials: Accuracy of Solution X-Ray Spectrometric Analysis of Copper-Base Alloys, March 1965. 25 cents.* (Out of print).
- NBS Misc. Publ. 260-6, Standard Reference Materials: Methods for the Chemical Analysis of White Cast Iron Standards, July 1965. 45 cents.*
- NBS Misc. Publ. 260-7, Standard Reference Materials: Methods for the Chemical Analysis of NBS Copper-Base Spectrochemical Standards, October 1965. 60 cents.*
- NBS Misc. Publ. 260-8, Standard Reference Materials: Analysis of Uranium Concentrates at the National Bureau of Standards, December 1965. 60 cents.* (Out of print).
- NBS Misc. Publ. 260-9, Standard Reference Materials: Half Lives of Materials Used in the Preparation of Standard Reference Materials of Nineteen Radioactive Nuclides Issued by the National Bureau of Standards, November 1965. 15 cents.*
- NBS Misc. Publ. 260-10, Standard Reference Materials: Homogeneity Characterization on NBS Spectrometric Standards II: Cartridge Brass and Low-Alloy Steel, December 1965. 30 cents.*
- NBS Misc. Publ. 260-11, Standard Reference Materials: Viscosity of a Standard Lead-Silica Glass, November 1966. 25 cents.*
- NBS Misc. Publ. 260-12, Standard Reference Materials: Homogeneity Characterization of NBS Spectrometric Standards III: White Cast Iron and Stainless Steel Powder Compact, September 1966. 20 cents.*
- NBS Misc. Publ. 260-13, Standard Reference Materials: Mossbauer Spectroscopy Standard for the Chemical Shift of Iron Compounds, July 1967. 40 cents.*
- NBS Misc. Publ. 260-14, Standard Reference Materials: Determination of Oxygen in Ferrous Materials — SRM 1090, 1091, and 1092, September 1966. 30 cents.*
- NBS Misc. Publ. 260-15, Standard Reference Materials: Recommended Method of Use of Standard Light-Sensitive Paper for Calibrating Carbon Arcs Used in Testing Textiles for Colorfastness to Light, June 1967. 20 cents.*
- NBS Spec. Publ. 260-16, Standard Reference Materials: Homogeneity Characterization of NBS Spectrometric Standards IV: Preparation and Microprobe Characterization of W-20% Mo Alloy Fabricated by Powder Metallurgical Methods, January 1969. 35 cents.*
- NBS Spec. Publ. 260-17, Standard Reference Materials: Boric Acid; Isotopic and Assay Standard Reference Materials, February 1970. 65 cents.*
- NBS Spec. Publ. 260-18, Standard Reference Materials: Calibration of NBS Secondary Standard Magnetic Tape (Computer Amplitude Reference) Using the Reference Tape Amplitude Measurement "Process A", November 1969. 50 cents.*
- NBS Spec. Publ. 260-19, Standard Reference Materials: Analysis of Interlaboratory Measurements on the Vapor Pressure of Gold (Certification of Standard Reference Material 745), January 1970. 30 cents.*
- NBS Spec. Publ. 260-20, Standard Reference Materials: Preparation and Analysis of Trace Element Glass Standards. (In preparation)
- NBS Spec. Publ. 260-21, Standard Reference Materials: Analysis of Interlaboratory Measurements on the Vapor Pressures of Cadmium and Silver, January 1971. 35 cents.*

- NBS Spec. Publ. 260-22, Standard Reference Materials: Homogeneity Characterization of Fe-3Si Alloy, February 1971. 35 cents.*
- NBS Spec. Publ. 260-23, Standard Reference Materials: Viscosity of a Standard Borosilicate Glass, December 1970. 25 cents.*
- NBS Spec. Publ. 260-24, Standard Reference Materials: Comparison of Redox Standards, January 1972. \$1.*
- NBS Spec. Publ. 260-25, Standard Reference Materials: A Standard Reference Material Containing Nominally Four Percent Austenite, February 1971. 30 cents.*
- NBS Spec. Publ. 260-26, Standard Reference Materials: National Bureau of Standards—U.S. Steel Corporation Joint Program for Determining Oxygen and Nitrogen in Steel, February 1971. 50 cents.*
- NBS Spec. Publ. 260-27, Standard Reference Materials: Uranium Isotopic Standard Reference Materials, April 1971. \$1.25.*
- NBS Spec. Publ. 260-28, Standard Reference Materials: Preparation and Evaluation of SRM's 481 and 482 Gold-Silver and Gold-Copper Alloys for Microanalysis, August 1971. \$1.*
- NBS Spec. Publ. 260-29, Standard Reference Materials: Calibration of NBS Secondary Standard Magnetic Tape (Computer Amplitude Reference) Using the Reference Tape Amplitude Measurement "Process A-Model 2", June 1971. 60 cents.*
- NBS Spec. Publ. 260-30, Standard Reference Materials: Standard Samples Issued in the USSR (A Translation from the Russian), June 1971. \$1.*
- NBS Spec. Publ. 260-31, Standard Reference Materials: Thermal Conductivity of Electrolytic Iron SRM 734 from 4 to 300 K, November 1971. 35 cents.*
- NBS Spec. Publ. 260-32, Standard Reference Materials: The Cooperative Study of Temperature Scale Standards for DTA by ICTA and NBS. (In preparation)
- NBS Spec. Publ. 260-33, Standard Reference Materials: Comparison of Original and Supplemental SRM 705, Narrow Molecular Weight Distribution Polystyrene, H. L. Wagner, May 1972. 35 cents.*
- NBS Spec. Publ. 260-34, Standard Reference Materials: Thermoelectric Voltage, April 1972. 40 cents.*
- NBS Spec. Publ. 260-35, Standard Reference Materials: Thermal Conductivity of Austenitic Stainless Steel, SRM 735 from 5 to 280 K, April 1972. 35 cents.*
- NBS Spec. Publ. 260-36, Standard Reference Materials: A Referee Method for the Determination of Calcium in Serum. SRM 915, May 1972. \$1.25.*
- NBS Spec. Publ. 260-37, Standard Reference Materials: Methods of Analysis of NBS Clay Standards, June 1972. 75 cents.*
- NBS Spec. Publ. 260-38, Standard Reference Materials: Preparation and Calibration of Standards of Spectral Specular Reflectance, May 1972. 60 cents.*
- NBS Spec. Publ. 260-39, Standard Reference Materials: The Eddy Current Decay Method for Resistivity Characterization of High-Purity Metals, May 1972. 55 cents.*
- NBS Spec. Publ. 260-40, Standard Reference Materials: Selection of Thermal Analysis Temperature Standards Through a Cooperative Study (SRM 758, 759, 760), August 1972. 65 cents.*
- NBS Spec. Publ. 260-41, Standard Reference Materials: Use of Standard Light-Sensitive Paper for Calibrating Carbon Arcs used in Testing Textiles for Colorfastness to Light, August 1972. 30 cents.*
- NBS Spec. Publ. 260-42, Standard Reference Materials: The Characterization of Linear Polyethylene, SRM 1475. (In preparation)
- NBS Spec. Publ. 260-43, Standard Reference Materials: Preparation and Homogeneity Characterization of an Austenitic Iron-Chromium-Nickel Alloy. (In preparation)

*Send order with remittance to: Superintendent of Documents, U.S. Government Printing Office, Washington, D.C. 20402. Remittance from foreign countries should include an additional one-fourth of the purchase price for postage.

STANDARD REFERENCE MATERIALS:

PREPARATION AND HOMOGENEITY CHARACTERIZATION OF
AN AUSTENITIC IRON-CHROMIUM-NICKEL ALLOY

H. Yakowitz, A. W. Ruff and R. E. Michaelis
Institute for Materials Research
National Bureau of Standards
Washington, D. C. 20234

An alloy of weight fraction 0.710 iron, 0.183 chromium and 0.107 nickel was characterized at the micrometer level of spatial resolution by means of electron probe microanalysis. This alloy, designated SRM 479, is of suitable homogeneity for use as a standard in microanalytical techniques. The coefficient of variation for each of the three elements is not more than 1.5% based on about 800 analyses involving five different specimens. There is no statistically significant variation in composition within specimens or from specimen to specimen. Electron probe microanalysis was carried out using different instrumental conditions and operators. SRM 479 is supplied as a disk about 4.6mm diameter by 1mm thick.

Key words: Austenitic stainless steel; electron probe microanalysis; Fe-Cr-Ni alloy; homogeneity testing; metallography; stacking fault energy; Standard Reference Material.

1. INTRODUCTION

The National Bureau of Standards maintains a continuing program to provide standards suitable for quantitative microanalytical methods such as electron probe microanalysis, spark source mass spectrometry, and laser probe analysis. Several standard materials have already been certified under this program [1-6].¹ This report describes the methods and results of homogeneity characterization of the first ternary alloy to be so certified - an austenitic Fe-Cr-Ni alloy designated SRM 479.

This alloy was produced at NBS as a research material for a project concerned with the measurement of stacking fault energies [7]. Those results indicated a high degree of solute homogeneity on a microscopic level. Additional thermomechanical treatment led to material of suitable homogeneity for use in microanalytical methods of analysis. The material preparation and the results of quantitative electron probe microanalysis of the constituent three elements will be presented in detail.

II. PREPARATION OF THE ALLOY

The alloy was melted by the arc fusion method using appropriate amounts of the raw material in the form of chips and small pieces. The raw material purities furnished by the suppliers were: Ni-99.98%, Cr-99.95%, and Fe-99.95%. An inert gas atmosphere of argon, gettered by titanium, was established within the furnace at a pressure of about 0.4 atmospheres. A non-consumable thoriated tungsten electrode was used to melt the charge of approximately 55g. The ingot was melted four times, twice on each side, and allowed to solidify in a round depression in the cooled, copper base plate. A final remelting was conducted in a well of rectangular cross-section in the base plate to produce an ingot about 15cm x 1cm x 1cm. The expected composition,

¹Figures in brackets indicate the references at the end of this paper.

assuming no material loss, was 10.6% Ni, 18.8% Cr, 70.6% Fe.

A portion of the ingot was reserved for rolling thin foil, a small amount for chemical analysis, and the remainder fabricated for the present standard. Initially, the balance of the ingot was annealed for 1 hr. at 1120°C within a sealed quartz tube containing helium at a pressure of 0.3 atmospheres. The ingot was then swaged to about 60% reduction in diameter, sealed again in quartz and homogenized for 5 days at 1120°C. Two subsequent swaging and annealing steps produced the final round rod of 4.6mm diameter x 220mm long. On each occasion the ends were cropped and the surface etched deeply in a 20% HNO_3 \pm 3% HF solution. No lubrication was used during the mechanical processing. Each heating treatment was concluded by quenching the sealed ingot into water. Five disks, 1mm thick, were cut from the final rod at each end and at the 1/4, 1/2, and 3/4 positions. These disks were used for electron microprobe characterization of the material. Chemical analysis performed on a representative portion of the ingot indicated a composition of 10.7% Ni, 18.3% Cr, 0.012% C, the balance Fe.

SRM 479 is issued as a disk, approximately 1mm thick x 4.6mm diameter, in the as-cut condition after electrical discharge slicing. A typical microstructure found on a metallographically prepared and etched transverse surface is shown in Fig. 1. The as-deformed grain structure is evident. Small pits and voids that were not completely removed in the processing treatments are to be seen in the section. Inclusions were not detected in the specimens examined.

III. PREPARATION FOR MICROPROBE ANALYSIS

This material was easily prepared for microprobe examination. Mounted specimens were ground on water-lubricated SiC papers in the usual progression 80, 220, 400 and 600 grit size. Rough polishing was done on a canvas cloth

impregnated with 6 μ m diamond particles. Fine polishing was done on a Selvyt cloth impregnated with 1/4 μ m diamond particles. Both rough and fine polishing were carried out with the wheels rotating at 125 RPM. The specimen was finished using a Gamal cloth impregnated with 0.05 μ m Al₂O₃. Wheel rotation was 150 RPM. Total time was about 30 minutes to complete specimen preparation. Unetched specimens were characterized and analyzed with the electron probe microanalyzer.

IV. HOMOGENEITY CHARACTERIZATION

Five specimens were taken from the rod; one from each end and one from the 1/4, 1/2 and 3/4 positions. The specimens were tested for homogeneity to determine whether any significant composition gradients were present in both the transverse and longitudinal directions of the rod.

Homogeneity testing was carried out with a microprobe operating voltage of 20kV. The K α radiation from all three elements was monitored by spectrometers equipped with LiF crystals and sealed proportional counters. The x-ray emergence angle was 52.5°. Each element peak was found by scanning the peak profile by hand. Monitor current was adjusted to give about 25,000 counts per second on pure iron. Under these conditions, a 40 second counting interval on the specimen gave about 900,000 Fe counts, 340,000 Cr counts and 270,000 Ni counts.

The homogeneity test on each transverse specimen face consisted of collecting data along two diameters at right angles to one another and four chords of the specimen; points were spaced 200 μ m apart. The chord paths were one-fourth of the specimen diameter and parallel to the diameters discussed above. The procedure was repeated twice on different days with different orientation of the specimen. Approximately 160 points total were taken on each specimen.

In evaluating the results, the coefficient of variation in percent, CV, was taken to be the most useful indicator of homogeneity. The value of CV is defined as

$$CV = \frac{100s}{N} (\%)$$

where s is the standard deviation, in counts, for a particular data array (s is obtained in the usual fashion) and N is the average total number of counts.

In no case was CV greater than 1.5% for any element present. Therefore, a conservative estimate of CV for the entire lot is 1.5% for all three elements, iron, chromium and nickel.

On any given day of testing, at least two different specimens were tested under the same experimental conditions. These were checked by an "outside count" method [8] to see whether, statistically, the two specimens were significantly different. The specimens always passed the test. Hence, we conclude that all five specimens tested are not statistically different from one another in terms of composition.

The implications of the CV value of 1.5% must now be considered. The width of a 99% confidence interval is

$$W_{99} = \pm C_i \frac{t_{n-1}^{99}}{\sqrt{n}} \frac{(CV)}{100} \quad (1)$$

where C_i is the true chemical weight fraction of element i in SRM 479; CV is the coefficient of variation, 1.5% in this case; n is the number of trials or analyses; and t_{n-1}^{99} is the Student's t value for 99% confidence for (n-1) degrees of freedom.

If "n" is arbitrarily made equal to sixteen in Eqn. (1), then

$$W_{99} = \pm C_i \frac{2.947}{4} (0.015) = \pm 0.0111 C_i \quad (2)$$

Hence, if one is willing to run sixteen separate points for each element in SRM 479, the 99% confidence interval widths and mean values would be:

$$\text{Iron} = 0.710 \pm 0.0080$$

$$\text{Chromium} = 0.183 \pm 0.0020$$

$$\text{Nickel} = 0.107 \pm 0.0012$$

The time to collect data for sixteen points is not prohibitive, being about 30 minutes.

V. ELECTRON PROBE MICROANALYSIS

As a final test, four different microprobe operators analyzed the specimens by means of electron probe microanalysis using pure iron, chromium and nickel as standards. The 1/4, 1/2, 3/4 specimens were used for analyses. The first of these was tested by analyst A, the second by B and the third by C. Analyst D collected data from all three specimens. Sixteen determinations were made for all elements. Results were obtained with instrumental operating voltages of 15, 20, 25, and 30kV respectively. Count rates were set so that pure iron gave about 12,000 counts per second to minimize coincidence loss uncertainties [9]. Instrument stability was checked by observing monitor current variations during the course of the analysis. Drift was less than one percent for all runs. Data from standards were obtained both before and after data from the specimen. No significant variation of the data from standards was observed.

Raw data from standards and specimen were entered in the microprobe data reduction program COR. This program has been described in detail elsewhere [10]. The results obtained are listed in Table I. These results show no significant specimen-to-specimen variations and reaffirm that the CV of 1.5% is reasonable. The analyses for iron and nickel are within about 1% relative for all operating voltages. The chromium analyses tend to be high, the most likely cause being the large characteristic fluorescence correction for chromium - about 20%. Errors in this correction have been discussed in detail by Heinrich and Yakowitz [11]. The fluorescence yield factors for nickel and iron may be the source of error in the case of the analyses in Table I.

VI. CONCLUSIONS

The ternary alloy consisting of 0.710 iron, 0.183 chromium and 0.107 nickel as determined by chemical analysis, is suitable for use as a microanalytical standard. This alloy, SRM 479, has a chemical homogeneity, expressed as coefficient of variation of not more than 1.5% for each element present. The individual region for which this coefficient of variation is valid is approximately a $1.5\mu\text{m}$ diameter sphere of the material. As an additional characterization, the stacking fault energy of this alloy has been determined from electron transparent thin foils in the temperature range 25°C to 325°C by Latanision and Ruff [7].

SRM 479 should be useful in the study of the fluorescence correction in microprobe analyses. This material will also be useful in any laboratory concerned with the quantitative microanalysis of stainless steels. SRM 479 represents the first ternary alloy to be certified under the homogeneity characterization program of the NBS Office of Standard Reference Materials.

The authors thank Mr. D. P. Fickle for his efforts in the preparation of SRM 479; Dr. K. F. J. Heinrich for use of the electron microprobe; Mrs. M. M. Darr, Mr. C. E. Fiori and Mr. R. L. Myklebust for electron probe microanalysis of SRM 479. Thanks are also due to Mr. Myklebust for aid in preparing the data for the computer. We thank Mr. R. A. Paulson for the chemical analysis of SRM 479. Finally, we acknowledge the initial characterization efforts of Dr. R. M. Latanision (RIAS) as part of his NAS-NRC Postdoctoral Research Associateship.

TABLE I. Electron Probe Microanalysis Results for Fe, Cr and Ni in SRM 479. Standards were pure Fe, Cr and Ni. True Concentrations in weight fractions: Fe = 0.710; Cr = 0.183; Ni = 0.107.

<u>Element</u>	<u>Voltage (kV)</u>	<u>k*</u>	<u>Calculated Concentration</u>	<u>CV (%) (16 points)</u>	<u>Analyst</u>
Fe	15	0.702	0.708	0.87	A
Fe	20	0.688	0.701	0.60	B
Fe	25	0.696	0.717	0.50	C
Fe	30	0.680	0.711	0.58	D
Cr	15	0.210	0.184	0.50	A
Cr	20	0.221	0.192	1.23	B
Cr	25	0.224	0.192	1.32	C
Cr	30	0.226	0.193	0.97	D
Ni	15	0.105	0.109	1.26	A
Ni	20	0.099	0.107	1.37	B
Ni	25	0.095	0.107	0.70	C
Ni	30	0.091	0.107	1.15	D

*The value of k is defined as the background-coincidence loss corrected ratio of intensities between the unknown and standard.

VII. REFERENCES

- [1] Michaelis, R. E., Yakowitz, H. and Moore, G. A., J. Res., Nat. Bur. Stand. (U.S.) 68A, 343 (1964).
- [2] Yakowitz, H., Vieth, D. L., Heinrich, K. F. J. and Michaelis, R. E., Advances in X-ray Analysis 9, 289 (1966).
- [3] Yakowitz, H., Vieth, D. L. and Michaelis, R. E., Nat. Bur. Stand. (U.S.) Spec. Publ. 260-12 (1966).
- [4] Yakowitz, H., Michaelis, R. E. and Vieth, D. L., Advances in X-ray Analysis 12, 418 (1969).
- [5] Yakowitz, H., Fiori, C. E. and Michaelis, R. E., Nat. Bur. Stand. (U.S.) Spec. Publ. 260-22 (1971).
- [6] Heinrich, K. F. J., Myklebust, R. L. and Rasberry, S. D., Nat. Bur. Stand. (U.S.) Spec. Publ. 260-28 (1971).
- [7] Latanision, R. M. and Ruff, A. W., Jr., Met. Trans. 2, 505 (1971).
- [8] Tukey, J. W., Technometrics 1, 31 (1959).
- [9] Heinrich, K. F. J., Vieth, D. L. and Yakowitz, H., Advances in X-ray Analysis 9, 208 (1966).
- [10] Heinrich, K. F. J., Henoc, J. and Myklebust, R. L., NBS Tech Note (1973), (in preparation).
- [11] Heinrich, K. F. J. and Yakowitz, H., Mikrochim. Acta 5, 905 (1968).

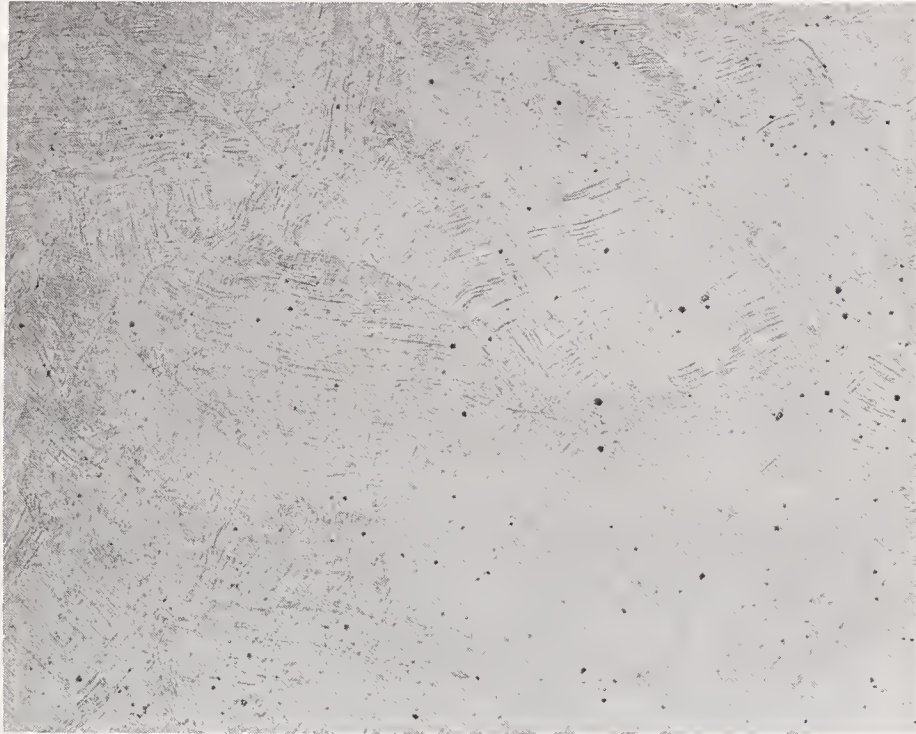
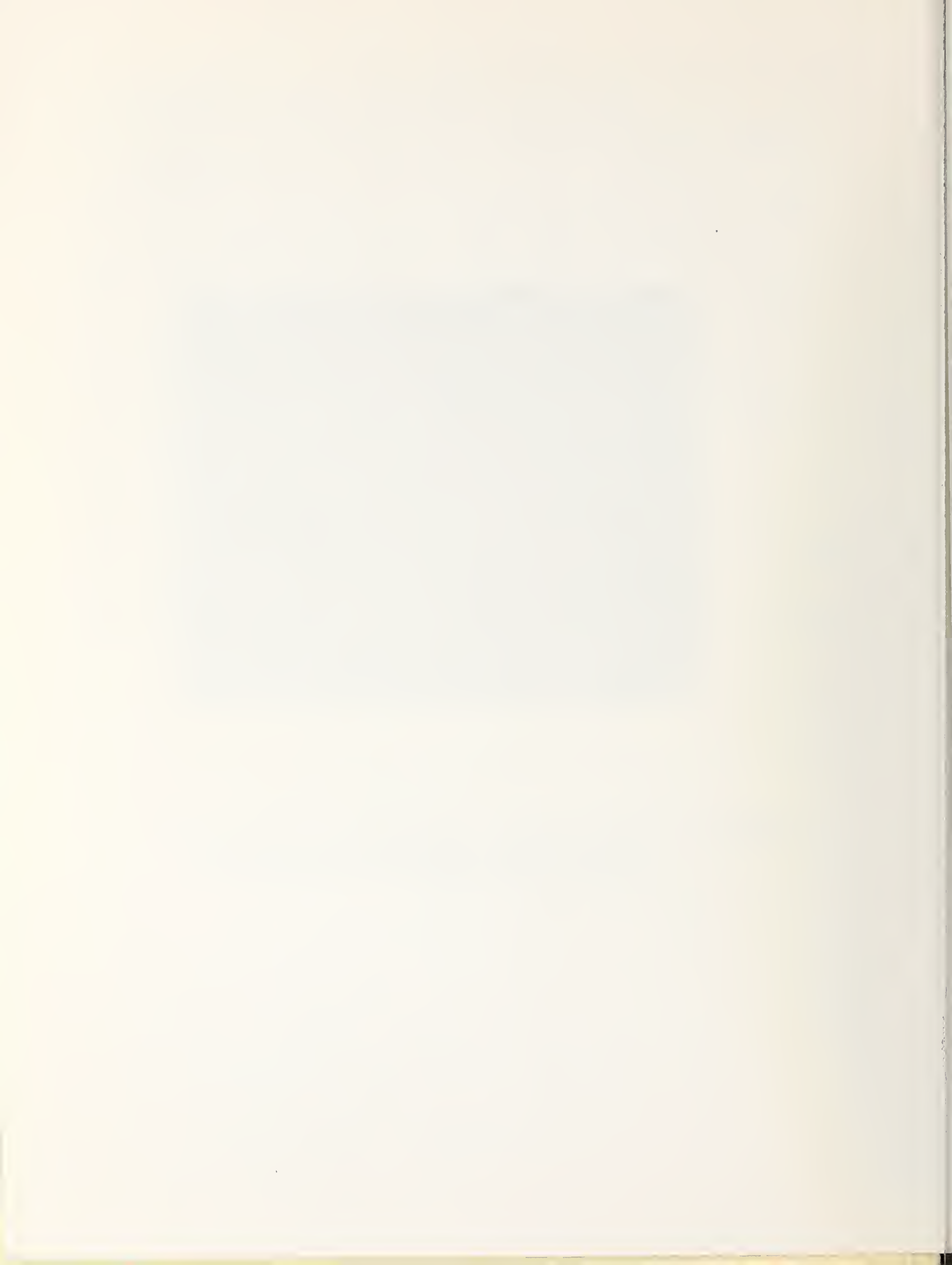


FIGURE I. Microstructure observed on a transverse surface after polishing and etching. Magnification 160X.



U.S. DEPT. OF COMM. BIBLIOGRAPHIC DATA SHEET		1. PUBLICATION OR REPORT NO. NBS SP 260-43	2. Gov't Accession No.	3. Recipient's Accession No.
4. TITLE AND SUBTITLE STANDARD REFERENCE MATERIALS: Preparation and Homogeneity Characterization of an Austenitic Iron-Chromium-Nickel Alloy			5. Publication Date November 1972	
			6. Performing Organization Code	
7. AUTHOR(S) H. Yakowitz, A. W. Ruff, R. E. Michaelis			8. Performing Organization	
9. PERFORMING ORGANIZATION NAME AND ADDRESS NATIONAL BUREAU OF STANDARDS DEPARTMENT OF COMMERCE WASHINGTON, D.C. 20234			10. Project/Task/Work Unit No. 3020101	
			11. Contract/Grant No.	
12. Sponsoring Organization Name and Address Same as No. 9.			13. Type of Report & Period Covered Final	
			14. Sponsoring Agency Code	
15. SUPPLEMENTARY NOTES				
16. ABSTRACT (A 200-word or less factual summary of most significant information. If document includes a significant bibliography or literature survey, mention it here.) An alloy of weight fraction 0.710 iron, 0.183 chromium and 0.107 nickel was characterized at the micrometer level of spatial resolution by means of electron probe microanalysis. This alloy, designated SRM 479, is of suitable homogeneity for use as a standard in microanalytical techniques. The coefficient of variation for each of the three elements is not more than 1.5% based on about 800 analyses involving five different specimens. There is no statistically significant variation in composition within specimens or from specimen to specimen. Electron probe microanalysis was carried out using different instrumental conditions and operators. SRM 479 is supplied as a disc about 4.6mm diameter by 1mm thick.				
17. KEY WORDS (Alphabetical order, separated by semicolons) Austenitic stainless steel; electron probe microanalysis; Fe-Cr-Ni alloy; homogeneity testing; metallography; stacking fault energy; Standard Reference Material.				
18. AVAILABILITY STATEMENT <input checked="" type="checkbox"/> UNLIMITED. <input type="checkbox"/> FOR OFFICIAL DISTRIBUTION. DO NOT RELEASE TO NTIS.			19. SECURITY CLASS (THIS REPORT) UNCLASSIFIED	21. NO. OF PAGES 19
			20. SECURITY CLASS (THIS PAGE) UNCLASSIFIED	22. Price 45 cents



PERIODICALS

JOURNAL OF RESEARCH reports National Bureau of Standards research and development in physics, mathematics, and chemistry. Comprehensive scientific papers give complete details of the work, including laboratory data, experimental procedures, and theoretical and mathematical analyses. Illustrated with photographs, drawings, and charts. Includes listings of other NBS papers as issued.

Published in two sections, available separately:

- **Physics and Chemistry**

Papers of interest primarily to scientists working in these fields. This section covers a broad range of physical and chemical research, with major emphasis on standards of physical measurement, fundamental constants, and properties of matter. Issued six times a year. Annual subscription: Domestic, \$9.50; \$2.25 additional for foreign mailing.

- **Mathematical Sciences**

Studies and compilations designed mainly for the mathematician and theoretical physicist. Topics in mathematical statistics, theory of experiment design, numerical analysis, theoretical physics and chemistry, logical design and programming of computers and computer systems. Short numerical tables. Issued quarterly. Annual subscription: Domestic, \$5.00; \$1.25 additional for foreign mailing.

TECHNICAL NEWS BULLETIN

The best single source of information concerning the Bureau's measurement, research, developmental, cooperative, and publication activities, this monthly publication is designed for the industry-oriented individual whose daily work involves intimate contact with science and technology—for *engineers, chemists, physicists, research managers, product-development managers, and company executives*. Includes listing of all NBS papers as issued. Annual subscription: Domestic, \$3.00; \$1.00 additional for foreign mailing.

Bibliographic Subscription Services

The following current-awareness and literature-survey bibliographies are issued periodically by the Bureau: Cryogenic Data Center Current Awareness Service (weekly), Liquefied Natural Gas (quarterly), Superconducting Devices and Materials (quarterly), and Electromagnetic Metrology Current Awareness Service (monthly). Available only from NBS Boulder Laboratories. Ordering and cost information may be obtained from the Program Information Office, National Bureau of Standards, Boulder, Colorado 80302.

NONPERIODICALS

Applied Mathematics Series. Mathematical tables, manuals, and studies.

Building Science Series. Research results, test methods, and performance criteria of building materials, components, systems, and structures.

Handbooks. Recommended codes of engineering and industrial practice (including safety codes) developed in cooperation with interested industries, professional organizations, and regulatory bodies.

Special Publications. Proceedings of NBS conferences, bibliographies, annual reports, wall charts, pamphlets, etc.

Monographs. Major contributions to the technical literature on various subjects related to the Bureau's scientific and technical activities.

National Standard Reference Data Series. NSRDS provides quantitative data on the physical and chemical properties of materials, compiled from the world's literature and critically evaluated.

Product Standards. Provide requirements for sizes, types, quality, and methods for testing various industrial products. These standards are developed cooperatively with interested Government and industry groups and provide the basis for common understanding of product characteristics for both buyers and sellers. Their use is voluntary.

Technical Notes. This series consists of communications and reports (covering both other-agency and NBS-sponsored work) of limited or transitory interest.

Federal Information Processing Standards Publications. This series is the official publication within the Federal Government for information on standards adopted and promulgated under the Public Law 89-306, and Bureau of the Budget Circular A-86 entitled, Standardization of Data Elements and Codes in Data Systems.

Consumer Information Series. Practical information, based on NBS research and experience, covering areas of interest to the consumer. Easily understandable language and illustrations provide useful background knowledge for shopping in today's technological marketplace.

CATALOGS OF NBS PUBLICATIONS

NBS Special Publication 305, Publications of the NBS, 1966-1967. When ordering, include Catalog No. C13.10:305. Price \$2.00; 50 cents additional for foreign mailing.

NBS Special Publication 305, Supplement 1, Publications of the NBS, 1968-1969. When ordering, include Catalog No. C13.10:305/Suppl. 1. Price \$4.50; \$1.25 additional for foreign mailing.

NBS Special Publication 305, Supplement 2, Publications of the NBS, 1970. When ordering, include Catalog No. C13.10:305/Suppl. 2. Price \$3.25; 85 cents additional for foreign mailing.

U.S. DEPARTMENT OF COMMERCE
National Bureau of Standards
Washington, O.C. 20234

OFFICIAL BUSINESS

Penalty for Private Use, \$300

POSTAGE AND FEES PAID
U.S. DEPARTMENT OF COMMERCE
215

